Review

Received January 24, 2022 Accepted for Publication February 28, 2022 ©2022 Soc. Mater. Eng. Resour. Japan

Review of the Mechanism for Orfom® D8 Depression of Chalcopyrite in Cu-Mo Separation during Cleaner Flotation

Courtney Young¹, Avimanyu Das¹, Rick LaDouceur², Simon Timbillah³ and Shawn Childress⁴

Montana Technological University, Butte MT 59701 USA

¹ Department of Metallurgical and Materials Engineering, CYoung@mtech.edu and ADas@mtech.edu

² Department of Mechanical Engineering, RLaDouceur@mtech.edu

³ Freeport-McMoRan, Tucson AZ 85622 USA, Aloribatim@gmail.com

⁴ Chevron Phillips Chemical, Bartlesville OK 74003 USA, childrs@cpchem.com

For the past 5 years, Montana Tech has coordinated efforts with Chevron Phillips Chemical to determine how the organic depressant, disodium carboxymethyl trithiocarbonate (Orfom® D8 depressant), works to separate chalcopyrite from molybdenite during cleaner flotation of a bulk sulfide concentrate. Pure minerals were studied by cyclic voltammetry, X-ray photoelectron spectroscopy, Fourier transform infrared spectroscopy, zeta potential and adsorption density. Results illustrated that Orfom® D8 was specific for chalcopyrite by chemisorbing electrochemically through its thiol functionality (CSS) thereby allowing the carboxylate group (COO) to extend into solution and cause depression. Density Function Theory via molecular orbital modeling corroborated the results. Flotation results obtained with Orfom® D8 using laboratory simple kinetic tests (SKT), bench-scale experiments, and industrial plant trials revealed grade and recovery of both Cu and Mo were better than those obtained with the most common inorganic depressant, NaSH. Replacing NaSH with Orfom® D8 in Cu-Mo flotation is therefore recommended and would simultaneously address mining sustainability issues.

Keywords: Orfom® D8, depressant, chalcopyrite, molybdenite, flotation